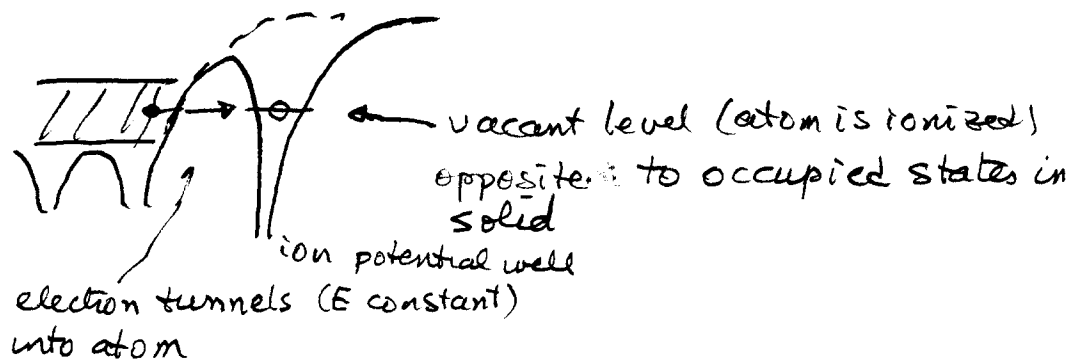
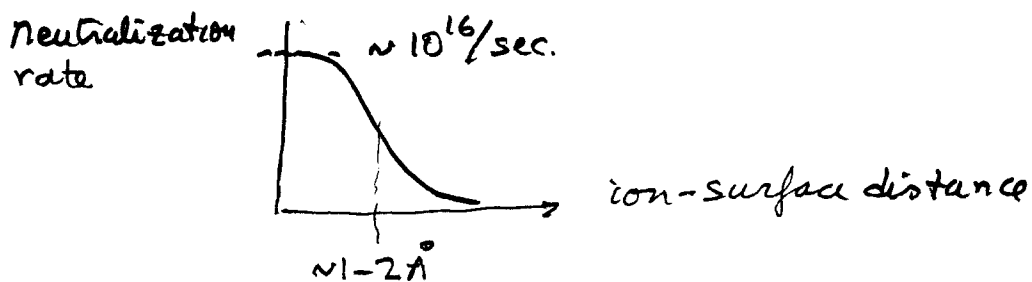


Electron Transfer at Surfaces

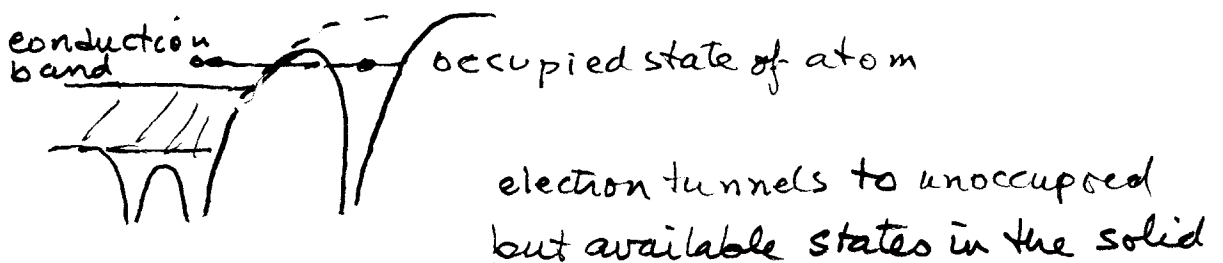
Resonant neutralization of + ion



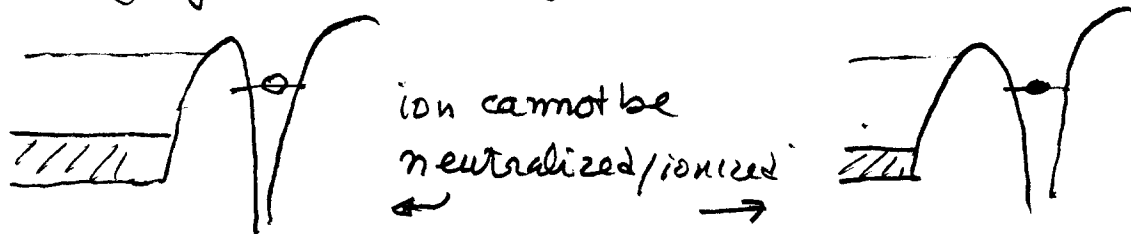
Electron tunnels through barrier



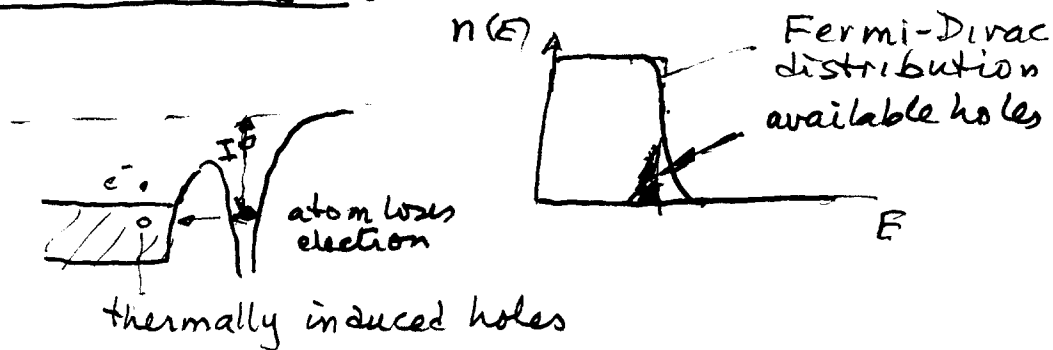
Resonant ionization of atoms or - ions



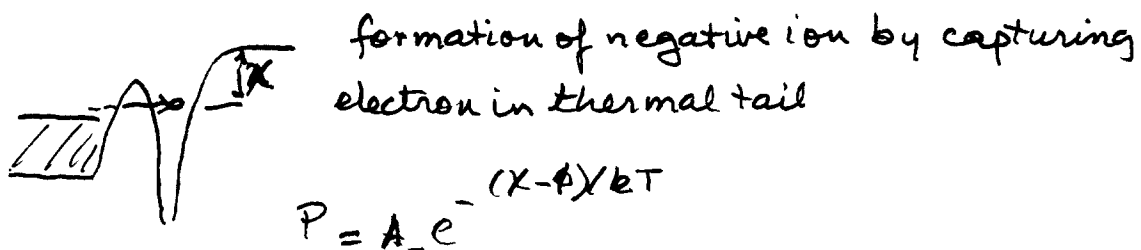
Blocking of electron transfer in insulators



Thermal ionization



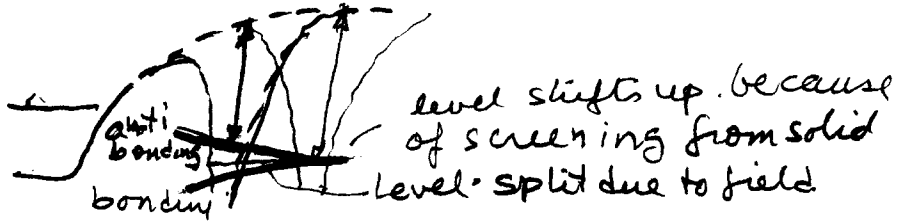
$$P_{\text{ioniz}} = A e^{-\frac{(IP - \phi)}{kT}}$$



in both cases one gets $\exp(-\Delta E/kT)$ where ΔE is the difference in energy between the atomic state and the Fermi level

Effect of image interaction - Level shift

Neutral atom



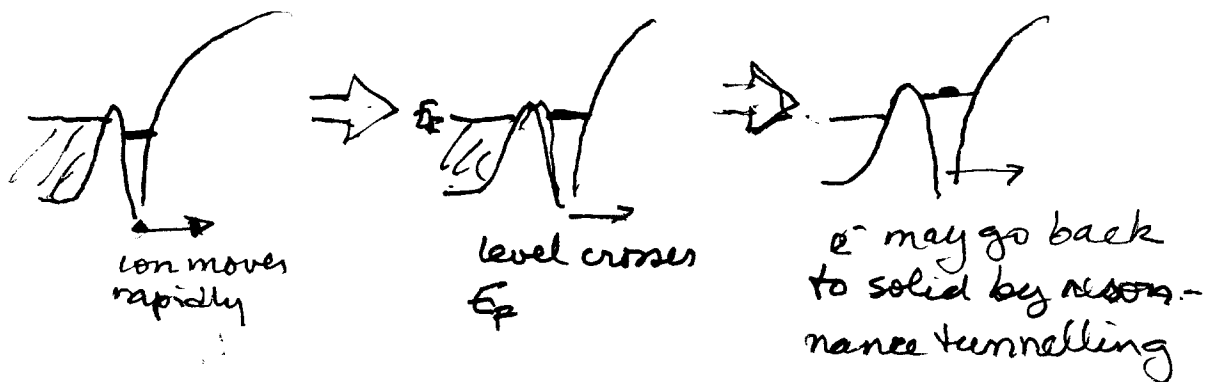
ionization potential decreases because vacuum level follows image interaction; Ionized electron is attracted to solid

Negative ion



extra e^- is attracted by image. Level sinks up to a few eV is possible.

Sputtering of negative ions



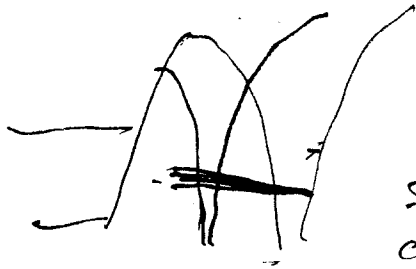
Survival depends on transit time near surface (on $v \perp$ surface)

Broadening

$$\Delta E = \frac{\hbar}{\tau}$$

(Heisenberg)

τ - lifetime of state (time to observe it)



state broadens when atom gets close to surface since when E is measured by removing electron, there is a fast re-neutralization from the solid